Koskinen, Ari M.P.; Brunner, Martin; Straub, Thomas; Saarenketo, Pauli; Rissanen, Kari

**Highly Diastereoselective Methylation of Five-Ring N,O-Acetals**

Published in: Letters in Organic Chemistry

Published: 01/01/2004

*Document Version*
Peer reviewed version

*Please cite the original version:*
Highly Diastereoselective Methylation of Five-Ring $N,O$-Acetals

Martin Brunner$^a$, Thomas Straub$^a$, Pauli Saarenketo$^b$, Kari Rissanen$^b$ and Ari M. P. Koskinen$^a$

$^a$Laboratory of Organic Chemistry, Helsinki University of Technology, P.O. Box 6100, FIN–02150 Espoo, Finland

$^b$X-ray Crystallography Laboratory of Organic Chemistry, Department of Chemistry, University of Jyväskylä, P.O. Box 35, FIN–40014 University of Jyväskylä, Finland

Abstract: Highly diastereoselective methylation of (2$S$,4$S$)- and (2$R$,4$S$)-3-tert-butyl 4-methyl 2-tert-butyloxazolidine-3,4-dicarboxylate (1a/b) is reported. The relative and absolute configuration of the methylated products was assigned by NOESY and confirmed by a crystal structure of 1a.

Keywords: Amino acids, alkylation, diastereoselectivity.

INTRODUCTION

In the course of our studies towards new synthetic routes for sphingosine-related metabolites [1] such as myriocin [2], mycestericins [3], and sphingofungins [4], we became interested in $\alpha$ substituted amino acids [5] (Fig. 1). We chose the principle of self-regeneration of stereocenters (SRS) [6] as the general synthetic strategy for these compounds. For these purposes, L-serine is a convenient starting material since it already contains all the functionalities of the hydrophilic end of the metabolites. For stereoselective $\alpha$ alkylation of this particular amino acid, a new stereogenic center is transiently introduced via formation of oxazolidines of type 1. Although similar oxazolidines have previously been prepared, we herein report new alkylation products. In contrast to recent literature, we have not only regenerated configuration at the defined stereocenter, but we have also prepared a product of inversion [6,7].

Fig. (1). Examples of Sphingosine-related Metabolites.

RESULTS AND DISCUSSION

The oxazolidines used in our alkylation studies were prepared according to standard methods (1a), or with a slightly modified literature procedure (1b) [7,8]. Compound 1a [9] was isolated from a 3:1 mixture together with 1b [10], while diastereopure 1b was obtained in three steps from L-serine (Scheme 1) [7]. Oxazolidines derived from serine esters are prone to ring-chain tautomerism [11].

Equilibrium studies of serine methyl and ethyl esters with aromatic aldehydes in CDCl$_3$ have shown three-component tautomeric mixtures where the open-chain Schiff base intermediate was typically predominating. Among the two ring forms the amount of the cis epimer was always higher than that of the trans epimer and, unlike the thiazolidines [12], no reaction conditions could be found to obtain predominantly the trans product [11]. We used the bulky N-protecting group Boc together with the large tBu ring substituent to obtain 1 in configurations where one face was maximally shielded from subsequent nucleophilic alkylation. In fact, 1a was in solution a 9:1 mixture of rotamers while only one rotameric form was observed for 1b [13].

Subsequent methylation reactions of the ester enolates of 1a and 1b were performed under standard conditions (Scheme 1) [14]. Using 110 mol-% of LDA was sufficient to deprotonate 1b but not 1a. However, the alkylation yields were only modest, mainly starting material being recovered, indicating slow enolization. Increasing the amount of base to 200 mol-% and prolonging the enolization time up to one hour improved the yield of the methylated products significantly. In contrast to literature, the use of DMPU as a co-solvent did not affect the outcome of the alkylations [7,15]. The methylated products 3a (87%) and 3b (92%) were obtained in high yield and purity [16]. Compounds 3a/b are enantiomers, which was confirmed by their identical NMR spectra and opposite signs of optical rotation.

The relative and thus also absolute configurations of 1a/b and 3a/b were determined with NOESY experiments.
and confirmed by the crystal structure of 1a. In the NOESY spectrum of 1b (not shown) there is a clear correlation between the methyl protons of the tBu and the acetal proton at C2 as well as the equatorial methylene proton at C5. The same correlations are observed for 3b, too (Fig. 2). Whereas there is no visible correlation between the pseudoaxial C5 proton and the C4 proton in 1b, we observed a correlation of the pseudoaxial methylene proton at C5 with the C4-methyl protons in 3b.

Based on the above arguments, the newly introduced substituent in 3b is necessarily trans to the tBu ring substituent. The relatively restricted rotational flexibility of 1a/b is corroborated by broader resonance signals of the acetal proton at C2 and the tBu/Boc methyl protons in the $^1$H NMR spectra both in CDCl$_3$ and benzene-$d_6$ at room temperature.

An ORTEP-3 [17] plot of 1a confirms the NOESY observations and the trans configuration of the C2 and C4 ring substituents, and illustrates how one face of the molecule ring is shielded by the large ring substituents (Fig. 3) [18].

In summary, selective formation of both diastereomers of 3-tert-butyl 4-methyl 2-tert-butyloxazolidine-3,4-dicarboxylate (1a/b) was achieved by two different routes, where the reversal of the order of protection steps is the main difference. Analytical data for 1a and 1b are presented for the first time and the observed structural and conformational properties were utilized to synthesise 3a and 3b in high purity, yield and dr. Further examples of this methodology and careful structural analysis of the highly substituted N,O-acetals will be reported in due course.

![Diagram of the synthesis and structures](image_url)

**Scheme 1.** Reagents and conditions: ia) (Boc)$_2$O, MeOH/CH$_2$Cl$_2$, r.t.; ib) (Boc)$_2$O, TEA/THF, r.t.; iia) toluene, reflux; iib) pivalaldehyde, TEA/pentane, reflux; iii) LDA, MeI, THF, -78 °C.

**Fig. (2).** NOESY spectrum of 3b.
ACKNOWLEDGEMENTS

This research was financially supported by the Ministry of Education (Graduate School of Bioorganic Chemistry Program) and the National Technology Agency (TEKES) of Finland.

REFERENCES


[9] (a) 1a: Rf = 0.17 (Hex/EtOAc 4:1). m.p. = 64-65 °C (Hex). [α]D = -57 (c 0.5, MeOH). 1H NMR (400 MHz, CDCl3), major rotamer, δ 5.17 (1H, br s), 4.33 (2H, m), 4.00 (1H, d, J = 7.2 Hz), 3.73 (3H, s), 1.43 (9H, br s), 0.93 (9H, s). 13C NMR (100 MHz, CDCl3), major rotamer, δ 172.0, 154.0, 96.8, 81.0, 74.0, 60.5, 52.3, 39.0, 28.1, 26.0. MS (EI) m/z: 288 (M+), 272, 232, 214, 200, 188, 174, 162, 146 (BP), 130, 128, 112, 102, 86, 70, 57. Calcd for C14H25NO5: C, 58.46; H, 8.77; N, 4.88. Found C, 58.04; H, 8.72; N, 4.66.

[10] 1b: Rf = 0.19 (Hex/EtOAc 4:1). [α]D = -30 (c 0.3, MeOH). 1H NMR (400 MHz, CDCl3) δ 5.00 (1H, s), 4.67 (1H, br m), 4.24 (1H, dd, J = 5.6, 8.6 Hz), 3.11 (1H, t, J = 8.6 Hz), 3.73 (3H, s), 1.45 (9H, s), 0.91 (9H, s). 13C NMR (100 MHz, CDCl3) δ 170.9, 151.5, 97.6, 81.3, 68.3, 59.7, 52.2, 37.7, 28.2, 25.8. HRMS calcd. for (C14H25NO5 + H) 288.1810, found 288.1823.


[13] For recent reviews, see: ref. [7b].

[14] General Procedure for Methylation: A 50 ml flask was charged with THF (35 ml) and diisopropylamine (0.1 ml, 0.70 mol). The solution was cooled to –78 °C and n-BuLi (0.70 mmol) was added. The reaction mixture was stirred at 0 °C for 15 min, then 1 (0.35 mol) in THF (5 ml) was added at –78 °C. Methyl iodide (0.88 mmol) was added after stirring for 45 min at –78 °C. The reaction mixture was warmed up to r.t. and then poured onto a mixture of semisaturated solution of NH4Cl (30 ml) and Et2O (80 ml). The organic layer was washed with dist. H2O (4 × 50 ml) and dried over MgSO4.


[16] (a) 3a: Rf = 0.16 (Hex/EtOAc 8:1). [α]D = +4 (c 0.4, CHCl3). 1H NMR (400 MHz, CDCl3) δ 5.10 (1H, s), 4.25 (1H, d, J = 8.0 Hz), 3.80 (1H, d, J = 8.0 Hz), 3.75 (3H, s), 1.40 (9H, s), 1.00 (9H, s). 13C NMR (100 MHz, CDCl3) δ 172.9, 151.4, 97.4, 81.2, 77.5, 66.8, 52.8, 39.6, 28.5, 26.9, 21.7. MS (EI) m/z: 302 (M+1) +, 296, 228, 214, 202, 186, 160 (BP), 114, 100, 84, 69, 58. HRMS calcd. for (C14H25NO5 + H) 302.1967, found 302.1977. 3b: [α]D = -5 (c 0.4, CHCl3).


[18] Crystal data for 1a: C14H25NO5, Mw = 287.35, orthorhombic, space group P212121, a = 5.9326(2), b = 10.1145(7), c = 26.3519(17) Å, α = 90°, β = 90°, γ = 90°, V = 107, Z = 4, T = 173(2) K, F(000) = 624, μ = 0.091 mm–1, 10166 reflections measured of which 3627 were independent, full matrix least-squares refinement on F2, R1 = 0.0931, wR2 = 0.2306. Data collection, 10166 reflections, 3627 independent (Rint = 0.1295); Full matrix least squares refinement on F2.