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Synthesis, in Vitro Evaluation, and Antileishmanial Activity of Water-Soluble Prodrugs of Buparvaquone

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Water-soluble phosphate prodrugs of buparvaquone (1), containing a hydroxynaphthoquinone structure, were synthesized and evaluated in vitro for improved topical and oral drug delivery against cutaneous and visceral leishmaniasis. The successful prodrug synthesis involved a strong base, e.g., sodium hydride. Buparvaquone-3-phosphate (4a) and 3-phosphonoxyethyl-buparvaquone (4b) prodrugs possessed significantly higher aqueous solubilities (>3.5 mg/mL) than the parent drug (<0.03 μg/mL) over a pH range of 3.0–7.4. Moreover, 4a and 4b maintained adequate lipophilicity as indicated by distribution coefficients (log D) between 0.5 and 3.0 over this pH range. Both 4a and 4b were also shown to be substrates for alkaline phosphatase in vitro and thus are promising bioreversible prodrugs for the improved topical and oral bioavailability of 1. Buparvaquone and its prodrugs showed nanomolar or low-micromolar ED₅₀ activity values against species that cause cutaneous leishmaniasis, e.g., L. major, L. amazonensis, L. aethiopica, L. mexicana, and L. panamensis and also L. donovani, which is the causative agent of visceral leishmaniasis. From these results, the human skin permeation of the prodrugs 4a and 4b were studied in vitro. While no buparvaquone permeated across post mortem skin in vitro during 72 h of experiments, both prodrugs 4a and 4b permeated readily through the skin. In addition, 4b easily released the parent drug in human skin homogenate and, therefore, is a promising prodrug candidate to deliver buparvaquone through the skin for the treatment of cutaneous leishmaniasis.

Introduction

Leishmaniasis is a widespread parasitic disease caused by protozoan parasites of the genus Leishmania. This disease occurs in two major forms: cutaneous (CL) and visceral leishmaniasis (VL) as well as rarer forms such as diffuse cutaneous (DCL) and mucocutaneous (MCL). The most severe form, VL, known as kala-azar, is nearly always fatal if not treated. Leishmaniasis is endemic in tropical and subtropical areas in both the Old and New Worlds. It is estimated that about 1.5 million new cases of CL and 0.5 million VL occur annually, and over 350 million people live at risk of infection. Over 20 species of Leishmania have been identified, and the most common species that cause CL in the Old World are L. major, L. tropica, and L. aethiopica and in the New World L. braziliensis, L. panamensis, L. amazonensis, and L. mexicana. Typically, CL presents as a nodule at the site of an insect bite, that gradually develops into an open lesion before self-cure within 3 to 18 months. VL is mainly caused by L. donovani, in which the parasite spreads from the site of inoculation to multiply as an amastigote in macrophages of the spleen, liver, lymph nodes, and bone marrow. Since the 1940s, pentavalent antimony compounds have been the main drugs used in antileishmanial chemotherapy. Although these drugs are effective, they have several disadvantages such as the need for parenteral administration at high dosages, long duration of therapy (several weeks), toxic side-effects, and variable efficacy. Thus, a real need exists for improved antileishmanials, particularly for oral administration in the treatment of VL and for topical administration in the treatment of CL.

The antiprotozoal activity of hydroxynaphthoquinones was already reported over 10 years ago. Many naphthoquinones have been tested against L. donovani, and other species both in vitro and in vivo, as oral, subcutaneous, and topical administrations. Buparvaquone has shown high activity in vitro against L. donovani amastigotes. However, the in vivo activity of buparvaquone against L. donovani was low after subcutaneous administration. This might be due to its poor distribution from the site of injection and subsequent lack of distribution of the drug to the intracellular target. In addition, the oral bioavailability of buparvaquone is low. These properties are most probably due to a combination of the low aqueous solubility and high lipophilicity of buparvaquone.

The prodrug approach is an effective means of improving the oral bioavailability of poorly water-soluble drugs by chemical derivatization to more water-soluble...
compounds. The most commonly used prodrug strategy for increasing aqueous solubility of drugs containing a hydroxyl group is to produce esters containing an ionizable group; e.g., phosphate group.\textsuperscript{9–11} Phosphate pro-moieties can be bound either directly or via a spacer group to molecules having an hydroxyl functionality.\textsuperscript{12,13} A prodrug is typically designed to release the active drug in the body during or after absorption by enzymatic and/or chemical hydrolysis. However, a phosphate group that is directly attached to a hydroxyl group is not always cleaved enzymatically due to steric hindrance. Therefore, much interest has been focused on designing enzymatically labile phosphonoxyxymethyl prodrugs.\textsuperscript{14–16} Such prodrugs release the parent drug via a two-step hydrolysis reaction; the first step is enzymatic cleavage of the ester group to give the hydroxymethyl intermediate, which then spontaneously releases the parent drug.

The prodrug approach is also widely used to improve topical drug delivery of various drugs by masking hydroxyl or amino groups via esterification.\textsuperscript{16–18} Topical formulations offer significant advantages over systemic therapy (e.g., antimonials) such as ease of administration, patient compliance, lower adverse effects, and low cost. Local treatment remains an attractive approach for the simple localized forms of CL, not at risk of MCL or involving critical organs. Treatment aims to accelerate healing and minimize scarring. A successful dermal prodrug should have an optimum lipophilicity (log D 2–3) and adequate aqueous solubility.\textsuperscript{19} In this study, phosphate prodrugs were synthesized to increase the aqueous solubility of buparvaquone and thus improve both oral and topical drug delivery properties. The hydrolytic properties of the prodrugs in aqueous buffers, specific enzyme solution, and human skin homogenate were determined in addition to in vitro permeation across post mortem human skin. The in vitro activities of buparvaquone and its phosphate prodrugs were also determined against the promastigote and amastigote forms of several Leishmania species causing both CL and VL.

### Results and Discussion

#### Synthesis

The prodrugs 4a and 4b were prepared by the route described in Scheme 1. Typical phosphorylation methods such as using phosphorus oxychloride\textsuperscript{22} or phosphorus pentoxide\textsuperscript{21} or synthesis with ditert-butyl-N,N-diisopropylphosphoramidite in the presence of 1H-tetrazole\textsuperscript{22} were unsuccessful. However, when 1 was first treated with sodium hydride followed by protected chlorophosphate (2a), 3a was produced in good yield. The ethyl protecting groups were removed by treatment with trimethylsilyl bromide in dry dichloromethane to yield the prodrug 4a. In the case of 4b, the protected chloromethyl phosphate 2b, prepared by an earlier described method,\textsuperscript{23} was condensed with 1 to produce 3b in moderate yield. After removal of the ditert-butyl protecting group by a saturated EtOAc/HCl solution, the resulting prodrug 4b was isolated. The preparation of the $-\text{O-CH}_2\text{-O-P-}$ functionality through a methylthiomethyl ether intermediate\textsuperscript{24,25} was not successful.

#### Aqueous Solubility, Lipophilicity, and Ionization Constant

The aqueous solubilities, distribution coefficients (log D) and ionization coefficients of 1 and the phosphate prodrugs 4a,b are summarized in Table.
The phosphate prodrugs would be sufficiently stable in chemical hydrolysis studies was to determine whether the subsequent precipitate in aqueous medium was not quantified due to its low aqueous solubility. In the present study, only the hydrolysis of prodrugs was determined. The degradation product (buparvaquone) was not quantified due to its low aqueous solubility. However, the subsequent precipitate in aqueous medium was verified to be buparvaquone. The aim of chemical hydrolysis studies was to determine whether the phosphate prodrugs would be sufficiently stable in

<table>
<thead>
<tr>
<th>Compd</th>
<th>pH 7.4</th>
<th>pH 5.0</th>
<th>pH 3.0</th>
<th>Log D</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>BPQ</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td>0.03 ± 0.01 μg/mL</td>
<td>&lt;0.03 μg/mL</td>
<td>&lt;0.03 μg/mL</td>
<td></td>
</tr>
<tr>
<td>4a</td>
<td>&gt;3.5(I)</td>
<td>&gt;3.5(I)</td>
<td>&gt;3.5(I)</td>
<td>0.47&lt;</td>
</tr>
<tr>
<td>4b</td>
<td>&gt;3.5(I)</td>
<td>&gt;3.5(I)</td>
<td>&gt;3.5(I)</td>
<td>1.16&lt;</td>
</tr>
</tbody>
</table>

* Mean ± SD; n = 3. * Determined using the capacity factors of RP-HPLC. * Determined by Sirius PCA 200. * Limited amount of the compound precluded further analyses.

Table 2. Rates of Hydrolysis of Buparvaquone Phosphate Prodrugs 4a,b in Buffer Solutions (pH 7.4, pH 5.0, and pH 3.0) at 37 °C

<table>
<thead>
<tr>
<th>Compd</th>
<th>t1/2 (h) (borate buffer pH 7.4)</th>
<th>t1/2 (h) (acetate buffer pH 5.0)</th>
<th>t1/2 (h) (citric acid buffer pH 3.0)</th>
</tr>
</thead>
<tbody>
<tr>
<td>4a</td>
<td>17</td>
<td>50</td>
<td>72</td>
</tr>
<tr>
<td>4b</td>
<td>0.5</td>
<td>8</td>
<td>97</td>
</tr>
</tbody>
</table>

Enzymatic Hydrolysis. Alkaline Phosphatase Solution. The phosphate esters 4a and 4b must undergo a chemical or enzymatic bioconversion to release the parent drug in order to fulfill the prodrg criteria. In addition to red blood cells alkaline phosphatases have been found in various human tissues, e.g., placenta, intestine, liver, bone, and kidney, and they have been reported to be the responsible enzymes for the hydrolysis of phosphate esters. Enzymatic hydrolysis was studied in alkaline phosphatase enzyme solution (Table 3). Both phosphate prodrugs 4a and 4b hydrolyzed rapidly (t1/2 = 72 and 97 h, respectively) for oral drug delivery (Table 2). For topical drug delivery, stabilities in aqueous solutions at pH 3–5 were low, and thus further formulation studies are needed to stabilize prodrugs 4a and 4b.

Enzymatic Hydrolysis. Skin Homogenate. The half-lives of 4a and 4b in human skin homogenate are summarized in Table 3. Similarly to chemical hydrolysis studies only hydrolysis of prodrugs was determined. The degradation product was confirmed to be buparvaquone. The rates of hydrolysis of 4a and 4b in human skin homogenate (pH 7.4) were from 2 to 3 times faster, respectively, than the rates of chemical hydrolysis in 0.185 M borate buffer (pH 7.4). These results suggest that the hydrolysis of the prodrugs in human skin homogenate is at least partly enzymatic hydrolysis. Because 4b hydrolyzed to buparvaquone rapidly in skin homogenate, it could be considered useful as a topical prodrg.

In Vitro Skin Permeation Study. The steady-state fluxes (Jss) of buparvaquone and prodrugs 4a and 4b are shown in Table 4. The results show that the post mortem skin permeation of buparvaquone can be improved by the phosphate prodrg strategy. While buparvaquone permeation in aqueous buffers or in buffer–ethanol mixtures was not observed at all, the prodrugs
activities in vitro against species that cause CL and thus using AlamarBlue. However, cultures should ideally be incubated for 48–72 h before final addition of the dye. In this assay, fluorescent readings were taken within 8 h post addition of the dye since fluorescent response did not increase proportionally with time during long incubation periods (>24 h). Fluorescence measurement, being more sensitive than the colometric measurement, was used in this previous paper, allowed shorter incubation with the dye.

The ED50 values for promastigotes indicated that buparvaquone and its prodrugs 4a and 4b have potent in vitro activity (2 to 124 nM), in many cases having greater activity than the amphotericin B control drug (Table 5). Against the majority of species investigated, the activity in order of decreasing potency is: buparvaquone > 4b > 4a > amphotericin B. The activity of prodrugs (4a and 4b) is probably due to their hydrolysis to buparvaquone during the experiment. Comparison of species sensitivity (Table 5) using the AlamarBlue method should be interpreted with caution. Differences in fluorescent response seen among species may result from rate/extent of AlamarBlue uptake or metabolism and cell number/size (reflected in sensitivity), rather than intrinsic sensitivity to drugs. After the 72-h incubation, none of the compounds (in media alone) were converted to the parent drug in human skin homogenate. Furthermore, the prodrug 4b was rapidly released in alkaline phosphatase solution.

The present study reports potent in vitro activity (nanomolar to low micromolar range) for buparvaquone and its phosphate prodrugs against the Leishmania species that cause CL and VL. Against the promastigote and amastigote form of most species investigated, the activity in order of decreasing potency was buparvaquone > 4b > 4a > control drug. It cannot be excluded that the activity caused by prodrugs (4a and 4b) is due to their conversion to buparvaquone. The study also shows that buparvaquone permeation through human skin can be significantly improved by using phosphate prodrugs. Furthermore, the prodrug 4b was rapidly converted to the parent drug in human skin homogenate. Thus, 3-phosphonoxyxymethyl-buparvaquone (4b) is a promising compound for topical treatment of CL. These phosphate prodrugs (4a and 4b) will be further investigated for both oral and topical drug administration in vivo.

### Conclusion

Buparvaquone-3-phosphate (4a) and 3-phosphonoxyxymethyl-buparvaquone (4b) were obtained in high yields by using sodium hydride. The aqueous solubility of buparvaquone was significantly increased by phosphorylation of the hydroxyl group and lipophilicity was maintained within an optimal range (log D 2–3) for absorption across biological membranes. Prodrugs 4a and 4b rapidly release buparvaquone in alkaline phosphatase solution.

### Experimental Section

**General Procedures.** 1H, 31P, and 13C NMR spectra were recorded on a Bruker Avance DRX500 spectrometer (Bruker, Rheinstetten, Germany) at 500.13, 202.45, and 125.76 MHz, respectively. TMS was used as an internal reference for 1H and 13C, and 85% H3PO4 as an external standard for 31P. The splitting pattern abbreviations are as follows: s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet, dt = doublet of triplets, bs = broad singlet, bm = broad multiplet. Mass spectra were acquired by a LCQ quadrupole ion trap mass spectrometer with an electrospray ionization source (Finnigan, San Jose, CA). The samples were dissolved in methanol (100 μg/mL) and 5 μL of samples were injected. The mobile phase consisted of acetonitrile/0.1% formic acid mixture (50:50 v/v), and the flow was set to 200 μL/min.

### Table 4. Steady-State Fluxes (Jss) for Delivery of Total Buparvaquone Species through Human Skin in Aqueous Buffers (pH 3.0 and pH 5.0) and Ethanol−Buffer (pH 5.0) Mixture at 37 °C

<table>
<thead>
<tr>
<th>compd</th>
<th>phosphate prodrugs (40 mM, pH 3.0)</th>
<th>phosphate prodrugs (50 mM, pH 5.0)</th>
<th>phosphate prodrugs (50 mM, pH 5.0)</th>
<th>and 50% ethanol</th>
</tr>
</thead>
<tbody>
<tr>
<td>BPQ</td>
<td>No drug permeation observed at 72 h.</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>4a</td>
<td>0.051 ± 0.02</td>
<td>0.133 ± 0.40</td>
<td>0.106 ± 0.08</td>
<td></td>
</tr>
<tr>
<td>4b</td>
<td>0.199 ± 0.01</td>
<td>1.500 ± 0.12</td>
<td>0.135 ± 0.05</td>
<td></td>
</tr>
</tbody>
</table>

* No drug permeation observed at 72 h.

4a and 4b permeated across the post mortem skin in all studied media. The prodrug 4b afforded a 11-fold and a 4-fold flux compared to 4a at pH 5.0 and pH 3.0, respectively, most probably due to its higher lipophilicity. A high drug concentration drives passive diffusion, and thus the prodrugs 4a and 4b themselves resulted in a 2.6-fold and a 7.5-fold higher fluxes at pH 5.0 than at pH 3.0, respectively. Moreover, buparvaquone and the phosphate prodrugs 4a and 4b have shown high activities in vitro against species that cause CL and thus may be good candidates for topical drug administration.

In Vitro Parasite Sensitivity. Assays involving both the extracellular promastigotes and intracellular amastigotes were carried out to confirm the antileishmanial activity of the compounds. The amastigote is the clinically relevant form of the parasite.

(a) Extracellular Promastigotes. Activity against L. major promastigotes has recently been determined using AlamarBlue. However, cultures should ideally be left to incubate for 48–72 h before final addition of the dye. In this assay, fluorescent readings were taken within 8 h post addition of the dye since fluorescent response did not increase proportionally with time during long incubation periods (>24 h). Fluorescence measurement, being more sensitive than the colometric assessment used in this previous paper, allowed shorter incubation with the dye.

The present study reports potent in vitro activity (nanomolar to low micromolar range) for buparvaquone and its phosphate prodrugs against the Leishmania species that cause CL and VL. Against the promastigote and amastigote form of most species investigated, the activity in order of decreasing potency was buparvaquone > 4b > 4a > control drug. It cannot be excluded that the activity caused by prodrugs (4a and 4b) is due to their conversion to buparvaquone. The study also shows that buparvaquone permeation through human skin can be significantly improved by using phosphate prodrugs. Furthermore, the prodrug 4b was rapidly converted to the parent drug in human skin homogenate. Thus, 3-phosphonoxyxymethyl-buparvaquone (4b) is a promising compound for topical treatment of CL. These phosphate prodrugs (4a and 4b) will be further investigated for both oral and topical drug administration in vivo.
TLC analyses were run on silica gel 60 F254 plates (Merck), and chromatography was performed with silica gel 60 (Merck) (0.063–0.200 mm). Elemental analyses were carried out on a ThermoQuest CE Instruments EA 110-CHNS-O elemental analyzer. All reagents were from commercial suppliers and used without further purification.

3-(trans-4-tert-Butyl-cyclohexylmethyl)-1,4-dioxo-1,4-dihydro-naphthalen-2-yl Diethyl Phosphosphate (3a). Compound 1 (2.0 g, 6.13 mmol) and 80% sodium hydride in mineral oil (0.37 g, 12.2 mmol) were dissolved in dry tetrahydrofuran (80 ml) under N2 atmosphere. The mixture was vigorously stirred at 0 °C for 30 min, followed by the addition of diethyl chlorophosphate (2a) (1.59 g, 9.2 mmol) in dry tetrahydrofuran (20 ml) with continuous vigorous stirring overnight at room temperature. The mixture was poured to ice (200 ml) and extracted three times with ethyl acetate (50 ml). The combined organic phase was dried with sodium sulfate and filtered, the solvent was evaporated, and the residue was dried under high vacuum. The product was purified by flash chromatography using petroleum ether/ethyl acetate (2:1) as eluent affording 3a as a yellow wax (2.36 g, 83.2%). TLC Rf 0.47 (petroleum ether/ethyl acetate 2:1). 1H NMR (CDCl3) δ 0.80 (9H, s, CMe3), 0.82–1.04 (5H, m, c-Hex), 1.44 (4H, t, J = 7.1 Hz, H2CH2(CH3)), 1.53 (1H, bm, c-HexH), 1.75 (4H, bm, c-HexH), 2.61 (2H, d, JHexH = 7.5 Hz, =CCH2C), 4.40 (4H, m, OCH2), 7.73 (2H, m, ArH), 8.08–8.12 (2H, m, ArH).

Mono-[3-(trans-4-tert-butyl-cyclohexylmethyl)-1,4-dioxo-1,4-di hydro-naphthalen-2-yl] Phosphate (4a). 3a (2.36 g, 5.10 mmol) was dissolved in dry dichloromethane (15 ml), and the solution was cooled to 0 °C. Excess trimethylsilyl bromide (5 ml) was added dropwise, and the mixture was allowed to warm to room temperature with continuous stirring for 4 h, then evaporated to dryness under high vacuum. The product was recrystallized from methanol to yield 4a as a yellow solid (1.01 g, 40.6%); mp 121 °C. 1H NMR (d6-DMSO) δ 0.80 (9H, s, CMe3), 0.82–1.02 (5H, m, c-Hex), 1.51 (1H, m, c-HexH), 1.67–1.73 (4H, bm, c-HexH), 2.54 (2H, d, J = 7.0 Hz, =CCH2C), 7.87 (2H, m, ArH), 8.02 (2H, m, ArH). 13C NMR (d6-DMSO) δ 26.7, 27.3, 31.0, 32.0, 33.2, 37.2, 71.25 (d, JCP = 1.4 Hz), 47.3, 125.8, 126.0, 130.5, 131.3, 134.0, 134.2, 135.7 (d, JCP = 6.0 Hz), 151.6 (d, JCP = 7.8 Hz), 178.8 (d, JCP = 0.9 Hz), 185.0 (d, JCP = 0.9 Hz). 31P NMR (CDCl3) δ −4.51 (s) ESI-MS: 405.6 (M − 1). Anal. (C37H20O10P1.0H2O) C, H.

Di-tert-butyl-3-(trans-4-tert-butyl-cyclohexylmethyl)-1,4-dioxo-1,4-di hydro-naphthalen-2-yl phosphonate (3b). Compound 1 (1.89 g, 5.80 mmol), 80% sodium hydride in mineral oil (0.28 g, 11.60 mmol), and n-BuLi (0.43 g, 11.6 mmol) were dissolved in dry dimethylformamide (DMF) (20 ml), and di-tert-butyl chloromethyl phosphate (2b) (1.50 g, 5.80 mmol) was added in DMF (10 ml) while stirring at 0 °C under N2 atmosphere. The mixture was allowed to warm to room temperature and stirred at 50 °C for 3 days. The reaction mixture was washed with saturated NaHCO3 solution and extracted three times with ethyl acetate. The combined organic layers were dried with sodium sulfate. The resulting residue was purified by flash column chromatography using petroleum ether/ethyl acetate (2:1) as eluent, affording 3b as a yellow oil (1.09 g, 34.3%). TLC Rf 0.37 (petroleum ether/ethyl acetate 2:1) 1H NMR (CDCl3) δ 0.80 (9H, s, CMe3), 0.82–1.06 (5H, m, c-HexH), 1.44 (18H, s, OCMes), 1.53 (1H, m, c-HexH), 1.67–1.73 (4H, bm, c-HexH), 2.54 (2H, d, J = 7.0 Hz, =CCH2C), 7.87 (2H, m, ArH), 8.02 (2H, m, ArH). 13C NMR (d6-DMSO) δ 26.7, 27.3, 31.0, 32.0, 33.2, 37.2, 71.25 (d, JCP = 1.4 Hz), 47.3, 125.8, 126.0, 130.5, 131.3, 134.0, 134.2, 135.7 (d, JCP = 6.0 Hz), 151.6 (d, JCP = 7.8 Hz), 178.8 (d, JCP = 0.9 Hz), 185.0 (d, JCP = 0.9 Hz). 31P NMR (CDCl3) δ −4.51 (s) ESI-MS: 405.6 (M − 1). Anal. (C37H20O10P1.0H2O) C, H.
c-HexH), 1.73 (4H, bm, c-HexH), 2.62 (2H, d, J = 7.3 Hz, =CCH=C), 5.82 (2H, d, J = 11.6 Hz, OCH3O), 7.70 (2H, m, ArH), 8.07 (2H, m, ArH).

Mono-[3-(trans-4-tert-buty1-cyclohexylmethyl)-1,4-dioxo-1,4-dihydro-naphthalen-2-ylmethyl] Phosphate (4b). A solution of 3b (0.5 g, 0.9 mmol) in ethyl acetate (2 mL) was added dropwise to saturated hydrochloric acid/ethyl acetate solution (10 mL) at 0 °C. The mixture was stirred at room temperature for 1 h and evaporated in vacuo. The residue was washed with ethyl acetate to yield 0.17 mg as a yellow solid (0.17 mg, 43.3%): mp 54 °C. 1H NMR (d6-DMSO) δ 0.79 (9H, s, CMe3), 0.82–1.02 (5H, m, c-HexH), 1.48 (1H, bm, c-HexH), 1.68 (4H, bm, c-HexH), 2.51 (2H, d, J = 7.0 Hz, =CCH=C), 5.70 (2H, d, J = 13.5 Hz, OCH3O), 7.84 (2H, m, ArH), 8.00 (2H, m, ArH). 13C NMR (d6-DMSO) δ 26.7, 27.3, 30.6, 32.0, 33.2, 37.2, 47.3, 89.9 (d, J = 4.6 Hz), 125.8, 125.9, 131.0, 131.4, 133.7, 134.1, 135.7, 180.2, 184.9. 31P NMR (CDCl3) δ −1.32 (s) ESI-MS: 435.5 (M–1). Anal. (C22H29O7P1,1/2) requires C, H.

HPLC Analysis. HPLC system consisted of a Beckman System Gold Programmable Solvent Module 126, a Beckman System Gold Detector Module 166 with variable wavelength (set at 250 nm for buparvaquone and 278 nm for its prodrugs) and a Beckman System Gold Autosampler 507e. Separations were performed with a Purospher RP18 column (125 × 4 mm, 5 μm). Gradient elution was carried out using mobile phase consisting of 90% (v/v) acetonitrile and 0.02 M phosphate buffer solution (pH 2.5) at a flow rate of 1.0 mL/min and at 30 °C.

Aqueous Solubility. Buparvaquone aqueous solubility was determined in phosphate buffer (0.16 M) at pH 3.0, pH 5.0, and pH 7.4 at room temperature. An excess of the compound was added to buffer solutions, and the suspensions were shaken for 72 h and filtered (Millipore 0.45 μm). The filtered solutions were analyzed by HPLC. The pH of the suspensions was held constant during the study. The aqueous solubilities of buparvaquone prodrugs 4a,b were determined in citric acid buffer (0.04 M) at pH 3.0, in sodium acetate buffer (0.05 M) at pH 5.0, and in borate buffer (0.185 M) at pH 7.4. The solutions were shaken for 1 h and filtered solutions were analyzed by HPLC.

Ionization Constant. Ionization constant (pK) titrations of the prodrugs were carried out by pH range of 1.5 to 10.5 under argon atmosphere at 25.2 °C (average temperature). A known amount of the compound (5.0–10.0 mg) was dissolved in three different methanol mixtures of 30, 40, and 50 (vol %) in 0.15 M KCl and consequently three separate titrations were performed. The cosomal, methanol, which normally is used to increase the solubility of compounds, was now used to stabilize the phosphate prodrugs during the measurements. The pH was adjusted automatically to pH 10.5, the starting pH of the titration. The maximum titrant volume increment for one titration step was limited to 0.25 mL. The pH change per titrant addition was limited to 0.2 pH units. A total of 42 pH readings were collected from each titration. The pH values were recorded when the pH-drift was lower than 0.002 pH per minute. Processing of the three separate titration data and the determination of the apparent pKs based on the Yasuda–Shedlovsky approach11 were carried out using the Sirius Refine200 Revision 1.000 (Sirius Analytical Instrument Ltd., Forest Row, UK). The used pH-metric method is described more detailed elsewhere.31,32

Distribution Coefficient. Distribution coefficients of the prodrugs were determined by the pH-metric technique using a Sirius PCA200 titrator. For the determination of distribution coefficient, 1-occtanol was saturated with 0.15 M KCl solution by vigorously shaking for 24 h. A known amount of the compound (5.0–10.0 mg) was dissolved in 7.5 mL of 0.15 M KCl solution, and three separate titrations were performed with different amounts of 1-occtanol (0.25 mL, 1.25 mL, 11.25 mL). The three separate measurements were combined in accordance with the MultiTitrations procedure.33 Processing of titration data was carried out using the Sirius Refine200 Revision 1.000. The used pH-metric log D method is described elsewhere.33 The log D value of buparvaquone was estimated from the capacity factors (k') of reversed-phase liquid chromatography.34 A Purospher RP–C8 column (125 × 4 mm, 5 μm) was used as the stationary phase, and the mobile phase consisted of 0.2 M phosphate buffer (pH 2.5) and methanol (30:70 v/v). The flow rate was 1.0 mL/min and the compounds were detected at 254 nm. The calibration curve was determined using nine standard compounds which are described elsewhere.35 The log D value of the prodrugs were plotted against the measured log k' values, and a linear relationship (r2 = 0.9469) was found between these two parameters. The log D value of buparvaquone was determined from this linear relationship, because the traditional shake-flask method was not useful due to the high lipophilicity and low aqueous solubility of 1.

Hydrolysis in Aqueous Solution. The chemical hydrolysis rates of 4a and 4b were determined in 0.04 M citric acid buffer (pH 3.0), 0.05 M sodium acetate buffer (pH 5.0), and 0.185 M borate buffer (pH 7.4) at 37 °C. An appropriate amount of prodrug was dissolved in 5 mL of preheated buffer, the filtered solutions (0.45 μm Millipore) were placed in a water bath at 37 °C, and at suitable intervals samples were withdrawn. The samples were then analyzed by HPLC. Pseudo-first-order half-life (t1/2) for the hydrolysis of prodrug was calculated from the slope of the linear portion of the plotted logarithm of remaining prodrug versus time.

Hydrolysis in Alkaline Phosphatase Solution. The rate of hydrolysis of prodrugs 4a and 4b in alkaline phosphatase solution was determined in 0.185 M borate buffer (pH 7.4) at 37 °C. An appropriate amount of 4a and 4b was added to 2 mL of preheated buffer solution and then filtered (0.45 μm Millipore). The solutions were placed in a water bath at 37 °C, and 2 μL (57.1 units) of alkaline phosphatase EC 3.1.3.1 (Type VII–S: from bovine intestinal mucosa, Sigma, St. Louis, MO) was added to the solution. Samples (200 μL) were withdrawn at determined intervals and 200 μL of cold acetonitrile was added to stop enzymatic hydrolysis. The samples were kept on ice for 1 h and centrifuged for 10 min at 14 000 rpm. The clear supernatant was analyzed for remaining prodrug by HPLC. Pseudo-first-order half-life (t1/2) for the hydrolysis of prodrug was calculated from the linear portion of the plotted logarithm of remaining prodrug against time.

Hydrolysis in Skin Homogenate. The rates of hydrolysis in post mortem human skin homogenate were studied by weighing an appropriate amount of the prodrug in 5.0 mL borate buffer (0.185 M, pH 7.4). The solutions were filtered (0.45 μm Millipore), and 0.5 mL of the preheated (37 °C) skin homogenate was added to 4.5 mL of the prodrug solution. The solutions were placed in a water bath and, at suitable intervals, 200 μL samples were withdrawn and 400 μL of cold acetonitrile was added to precipitate protein from the skin homogenate. After immediate mixing, the samples were kept on ice for 1 h, and then centrifuged for 10 min at 14 000 rpm. The supernatant was analyzed for remaining prodrug by HPLC using the method described earlier. Pseudo-first-order half-life (t1/2) for the hydrolysis of prodrug was calculated from the ratio of the line of the plotted logarithm of remaining prodrug against time. For the skin homogenate, a required amount of dried, post mortem human skin was homogenized (Ystral 10/25, dispensing tool 10G, Dottingen, Germany) in 0.185 M borate buffer (pH 7.4) at 22,000 rpm, to obtain a final concentration of 360 mg/mL. The supernatant was separated after centrifugation at 14,000 rpm (Eppendorf 5415C, Germany), and lipid drops were extracted from the surface of the clear mixture and then stored at −20 °C.

Vitro Skin Permeation Study. An epidermal membrane was obtained from the abdominal area of adult cadavers from the Kuopio University Hospital (Kuopio, Finland). Epidermal membranes were isolated from the dermis by heating at 60 °C in distilled water for 2 min, after which the skin specimens were dried 2–4 days at room temperature and frozen. The permeation studies were carried out in Franz-type diffusion cells (PermeGear, Inc., Riegelsville, PA). Skin specimens were
Promastigotes were plated at 1 × 10^5 macrophages/mL in RPMI 1640 supplemented with 10% HI-FCS; 100 μL added to each well of a 16-well tissue culture chamber slides (Lab-Tek, Life Technologies, Paisley, UK). After 24 h at 37 °C in 5% CO_2/95% air, macrophages were infected with 'stationary phase' Leishmania promastigotes at a ratio of 10 promastigotes: 1 macrophage and incubated at 34 °C. For L. donovani, amastigotes were used for infection at a ratio of 7 amastigotes: 1 macrophage and incubated at 37 °C. After 24 h, cultures were washed prior to addition of medium containing serial dilutions of the test compounds. Control slides were taken at this point to determined % infection. The infected cultures were exposed for 72 h to compounds in a 3-fold dilution series over four concentrations in quadruplicate at each concentration. After 72 h slides were fixed with methanol and stained with 10% (v/v) Giemsa's stain (BDH, UK). The % of infected macrophages was determined microscopically at each drug concentration, compared to the untreated 72-h control.

(e) Statistical Analysis. ED_50 values were estimated using Microsoft XLfit (version 2.0.2). Data were fitted using the nonlinear sigmoidal curve-fitting Levenburg Marquardt algorithm.

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